**9 Build Chemistry - Quick Quiz 2**

Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

**How much do you know and understand? Answer each question to the best of your ability.**

1. What are the three particles that make up an atom?
   1. Protons, neutrons, and isotopes
   2. Neutrons, isotopes, and electrons
   3. Positives, negatives, and electrons
   4. Protons, neutrons, and electrons
2. The space surrounding the nucleus of an atom contains\_\_\_\_\_\_\_\_\_\_\_

a. protons

b. electrons

c. neutrons

d. isotopes

1. Protons are:

a. positively charged

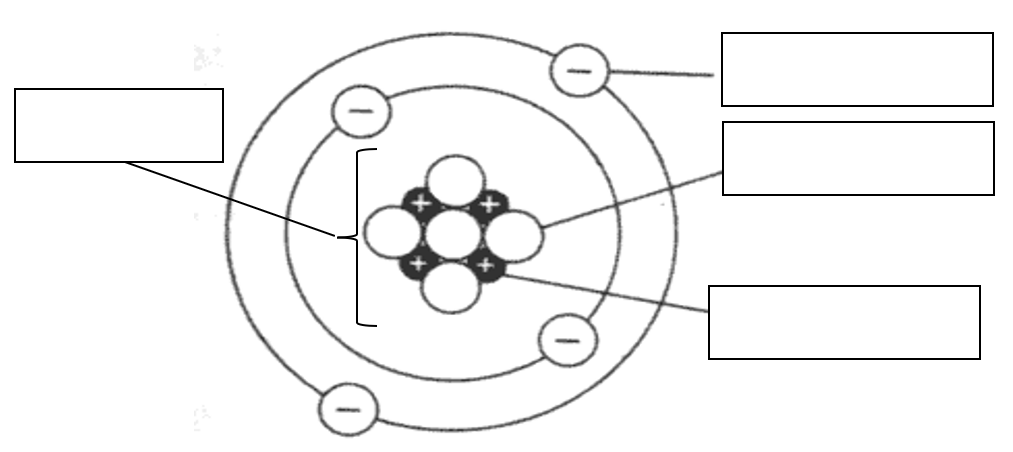
b. negatively charged

c. neutral

d. always equal to the number of neutrons in an atom

1. Label the parts of this atom:

|  |  |  |  |
| --- | --- | --- | --- |
| electron | proton | neutron | nucleus |



1. Name the radiation (alpha, beta or gamma) that best fits each of the following statements.

|  |  |
| --- | --- |
| Radiation (alpha, beta or gamma) | Statement |
|  | Stopped by thin paper or a few centimetres of air |
|  | The most massive radiation (heaviest) |
|  | Can only be stopped by lead or thick concrete |
|  | Can travel through paper but is stopped by aluminium |
|  | Has no mass and no charge |

1. Match the correct term with each definition

|  |  |  |
| --- | --- | --- |
| **Term** |  | **Definition** |
| Atomic Number |  | The smallest particles of an element that are the building blocks of matter. |
| Atom |  | The number of protons and neutrons in an atom. |
| Element |  | The number of protons in an atom |
| Mass Number |  | A pure substance consisting of only one type of atom. |

1. Complete the missing space(s) in each row of the following table:

|  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- |
| **Isotope** | **Atomic Number** | **Mass Number** | **Protons** | **Electrons** | **Neutrons** |
| Lithium - 7 | 3 | 7 |  | 3 | 4 |
| Carbon - 13 |  | 13 | 6 |  | 7 |
| Boron - 10 | 5 |  | 5 | 5 | 5 |
| Helium - 4 | 2 | 4 | 2 | 2 |  |